

One-pot Synthesis of Novel Polysubstituted Pyrazole and Pyrrolo[2,1-*b*]benzothiazole Derivatives†

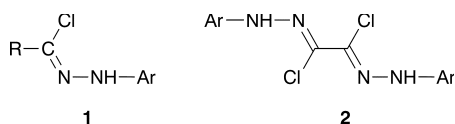
J. Chem. Research (S),
1998, 128–129†

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Reaction of hydrazonoyl chlorides **1** and bis-hydrazonoyl chlorides **2** with benzothiazole-2-acetonitrile furnished novel polysubstituted pyrazole and pyrrolo[2,1-*b*]benzothiazole derivatives.

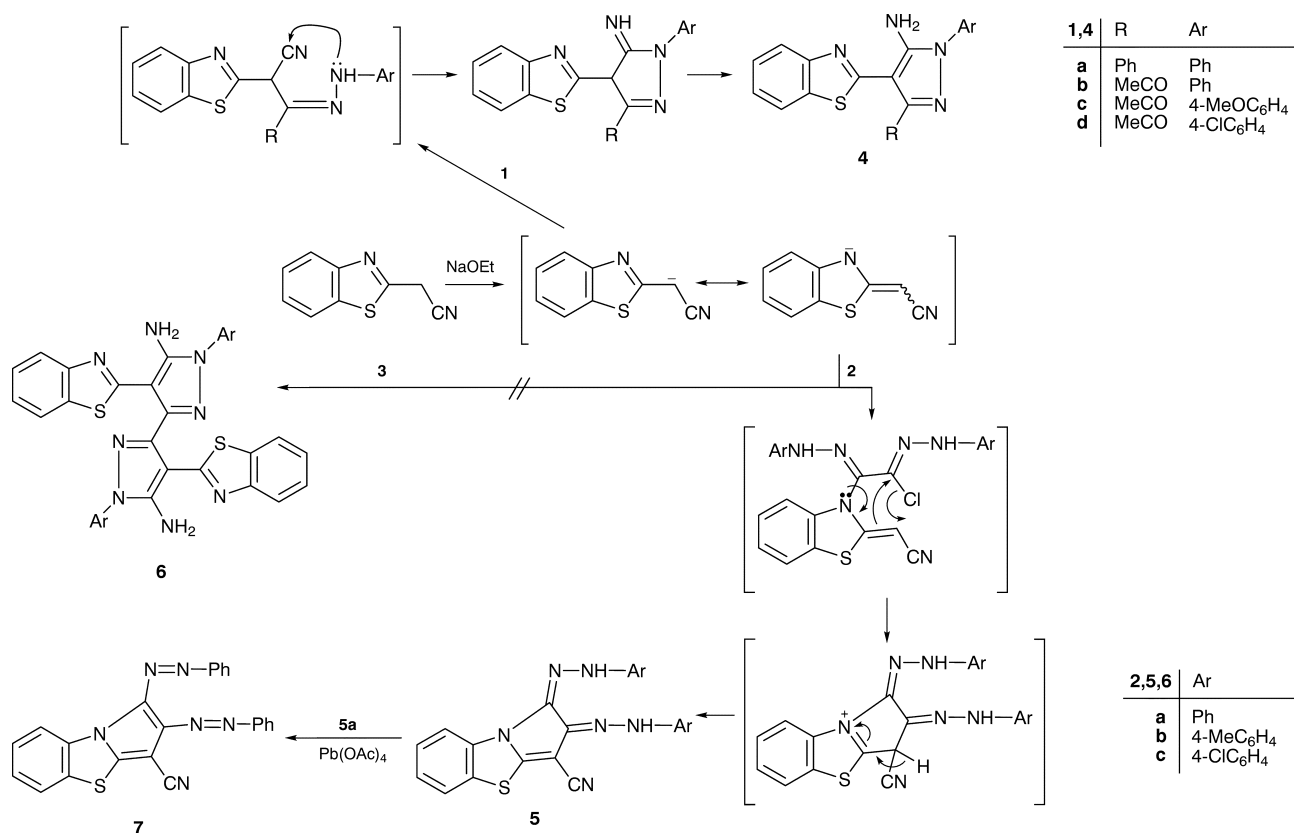
Hydrazonoyl halides are highly versatile intermediates for the synthesis of a variety of heterocyclic systems.¹ In continuation of our interest in the synthesis of novel poly-functionally substituted heterocycles^{2–6} of expected biological importance, we report here a facile, one-pot synthesis of the title compounds *via* reaction of hydrazonoyl chlorides **1** and bis-hydrazonoyl chlorides **2** with benzothiazole-2-acetonitrile.



Thus, treatment of the hydrazonoyl chlorides **1a–d** with benzothiazole-2-acetonitrile (**3**) in ethanolic sodium ethoxide solution, at room temperature, afforded products identified as 3-substituted 5-amino-1-aryl-4-(benzothiazol-2-yl)pyrazoles **4a–d** (Scheme 1). The appearance for each product of two characteristic IR absorption bands in the

region 3450–3200 cm^{-1} and a broad singlet in the NMR spectrum around δ 6.5 due to an amino function provided firm support for structure **4**.

However, when the bis-hydrazonoyl chloride **2a** was treated with 2 mol equivalents of compound **3** under the same reaction conditions, two products were isolated (as examined by TLC). Elemental analyses and spectral data proved that one of these is the unreacted benzothiazole-2-acetonitrile (**3**) (obtained from the mother-liquor after slight acidification). The main product obtained from the reaction mixture was found to be a 1:1 cycloaddition identified as 3-cyano-1,2-diphenylhydrazonopyrrolo[2,1-*b*]benzothiazole-2,3-dione (**5a**). The IR spectrum of **5a** exhibited two absorption bands at 3427 and 3270 cm^{-1} due to two hydrazone NH stretchings as well as a nitrile absorption band at 2218 cm^{-1} , while its ¹H NMR spectrum displayed two D₂O-exchangeable hydrazone NH protons at δ 11.62 besides an aromatic multiplet in the region δ 7.02–8.69. However, all attempts to prepare the 3,3'-bi(5-aminopyrazole)



Scheme 1

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†This is a **Short Paper** as defined in the Instructions for Authors, Section 5.0 [see *J. Chem. Research (S)*, 1998, Issue 1]; there is therefore no corresponding material in *J. Chem. Research (M)*.

derivative **6a** *via* reaction of **2a** with **3** in a 1:2 molar ratio were unsuccessful. A plausible mechanism for the formation of compounds **3** and **5** is depicted in Scheme 1. Furthermore, the bis(phenyl hydrazone) **5a** was substantiated by its

oxidation with lead tetraacetate in glacial acetic acid at reflux temperature to give 3-cyano-1,2-diphenylazopyrrolo[2,1-*b*]benzothiazole (7). Assignment of structure 7 was made on the basis of its spectral data where its IR spectrum was free of NH absorptions and exhibited only a nitrile absorption band at 2218 cm⁻¹. Prompted by these interesting results and in order to generalize this phenomenon, the reaction of compound 3 with other bis-hydrazonoyl chlorides 2b,c under the same experimental conditions was examined. The reaction products were found to be 1:1 cycloadducts identified as 3-cyano-1,2-diarylhydrazonopyrrolo[2,1-*b*]benzothiazole-1,2-diones 5b,c, respectively.

Experimental

Mps were measured on a Gallenkamp apparatus. IR spectra were recorded for KBr pellets on a Pye–Unicam SP 3-300 infrared spectrophotometer. ¹H NMR spectra were obtained in [2H]chloroform on a Varian Gemini 200 NMR spectrometer using tetramethylsilane as an internal standard. Mass spectra were recorded on a GCMS-QP 1000 EX mass spectrometer at 70 eV. Microanalyses were carried out at the Microanalytical Center, University of Cairo, Giza, Egypt.

Hydrazonoyl chlorides 1a⁷ and 1b–d,⁸ *N,N'*-diarylethanediohydrazonoyl chlorides 2a–c^{9,10} and benzothiazole-2-acetonitrile (3)¹¹ were prepared according to literature procedures.

*Synthesis of 5-Aminopyrazoles 4a–d and Pyrrolo[2,1-*b*]benzothiazoles 5a–c. General procedures.*—Benzothiazole-2-acetonitrile (3) (0.348 g, 2 mmol) was added to an ethanolic sodium ethoxide solution [prepared by dissolving sodium metal (46 mg, 2 mmol) in ethanol (25 ml)] with stirring. To the resulting solution was added the appropriate hydrazonoyl chloride 1a–d (2 mmol) or bis-hydrazonoyl chloride 2a–c (1 mmol) in portions and the mixture was left to stir overnight. The precipitated product was collected by filtration, washed with water, dried and finally recrystallized from ethanol or *N,N*-dimethylformamide–water to afford 4a–d and 5a–c in 53–78% and 65–74% yields, respectively. The mother-liquors of 5a–c were treated with water (10 ml) and concentrated hydrochloric acid (0.5 ml) and the solid product was filtered off and crystallized from ethanol to give, in each case, compound 3, mp 101–102 °C (lit.,¹¹ 101–102 °C). 4a (68%): mp 142–144 °C; $\nu_{\max}/\text{cm}^{-1}$ 3400, 3280 (NH₂), 1600 (C=N); δ_{H} 6.18 (2 H, s), 7.00–7.90 (14 H, m); m/z 368 (M⁺) (Found: C, 71.51; H, 4.19; N, 15.15; S, 8.73. C₂₂H₁₆N₄S requires C, 71.74; H, 4.34; N, 15.21; S, 8.69%). 4b (72%): mp 178–180 °C; $\nu_{\max}/\text{cm}^{-1}$ 3409, 3244 (NH₂), 1678 (C=O), 1608 (C=N); δ_{H} 2.68 (3 H, s), 6.52 (2 H, s), 7.26–7.90 (9 H, m); m/z 334 (M⁺) (Found: C, 64.70; H, 4.16; N, 16.54; S, 9.60. C₁₈H₁₄N₄OS requires C, 64.67; H, 4.19; N, 16.76; S, 9.58%). 4c (53%): mp 172–174 °C; $\nu_{\max}/\text{cm}^{-1}$ 3444, 3226 (NH₂), 1683 (C=O), 1606 (C=N); δ_{H} 2.68 (3 H, s), 3.88 (3 H, s), 6.42 (2 H, s), 7.05–7.91 (8 H, M); m/z 364 (M⁺) (Found: C, 62.65; H, 4.31; N, 15.29; S, 8.80. C₁₉H₁₆N₄O₂S requires C, 62.63; H, 4.39; N, 15.38; S, 8.79%).

4d (78%): mp 187–188 °C; $\nu_{\max}/\text{cm}^{-1}$ 3415, 3225 (NH₂), 1681 (C=O), 1608 (C=N); δ_{H} 2.68 (3 H, s), 6.53 (2 H, s), 7.20–7.98 (8 H, m); m/z 368 (M⁺) (Found: C, 58.58; H, 3.50; Cl, 9.55; N, 15.21; S, 8.70. C₁₈H₁₃ClN₄OS requires C, 58.61; H, 3.52; Cl, 9.63; N, 15.19; S, 8.68%). 5a (72%): mp 178–180 °C; $\nu_{\max}/\text{cm}^{-1}$ 3427, 3270 (2 NH), 2218 (C=N), 1596 (C=N); δ_{H} 7.02–8.69 (14 H, m), 11.62 (2 H, s); m/z 408 (M⁺) (Found: C, 67.58; H, 3.88; N, 20.42; S, 7.90. C₂₃H₁₆N₆S requires C, 67.64; H, 3.92; N, 20.58; S, 7.84%). 5b (65%): mp 184–186 °C; $\nu_{\max}/\text{cm}^{-1}$ 3446, 3295 (2 NH), 2214 (C=N), 1600 (C=N); δ_{H} 2.46 (6 H, s), 6.91–8.60 (12 H, m), 12.95 (2 H, s); m/z 436 (M⁺) (Found: C, 68.74; H, 4.50; N, 19.30; S, 7.23. C₂₅H₂₀N₆S requires C, 68.80; H, 4.58; N, 19.26; S, 7.34%). 5c (74%): mp 258–260 °C; $\nu_{\max}/\text{cm}^{-1}$ 3361, 3265 (2 NH), 2218 (C=N), 1600 (C=N); δ_{H} 7.10–8.64 (12 H, m), 12.60 (2 H, s); m/z 477 (M⁺) (Found: C, 57.80; H, 2.77; Cl, 14.82; N, 17.65; S, 6.67. C₂₃H₁₄Cl₂N₆S requires C, 57.86; H, 2.93; Cl, 14.88; N, 17.61; S, 6.70%).

Oxidation of 5a.—A mixture of 5a (0.408 g, 1 mmol) and lead tetraacetate (0.52 g, 1.2 mmol) in glacial acetic acid was refluxed for 2 h, then cooled. The solid product so formed was collected by filtration, washed with water and dried. Recrystallization from *N,N*-dimethylformamide afforded 3-cyano-1,2-bis(phenylazo)pyrrolo[2,1-*b*]benzothiazole (7) as reddish-brown crystals in 75% yield, mp 233–235 °C; $\nu_{\max}/\text{cm}^{-1}$ 2218 (C=N), 1625 (C=C); δ_{H} 7.06–8.55 (m, ArH's); m/z 406 (M⁺) (Found: C, 67.82; H, 3.39; N, 20.57; S, 7.90. C₂₃H₁₄N₆S requires C, 67.98; H, 3.44; N, 20.68; S, 7.88%).

Received, 20th August 1997; Accepted, 26th November 1997
Paper E/7/06107B

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